

K. Ravikumar,^{a*} B. Sridhar,^a M. Mahesh^b and V. V. Narayana Reddy^b^aLaboratory of X-ray Crystallography, Indian Institute of Chemical Technology, Hyderabad 500 007, India, and ^bOrganic Chemistry Division-II, Indian Institute of Chemical Technology, Hyderabad 500 007, IndiaCorrespondence e-mail:
ravikumar_iict@yahoo.co.in

Key indicators

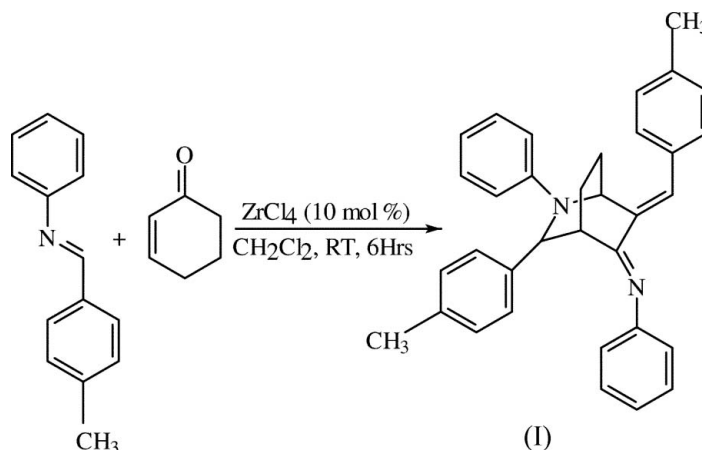
Single-crystal X-ray study
T = 273 K
Mean $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$
R factor = 0.043
wR factor = 0.127
Data-to-parameter ratio = 13.1For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.***N*-{3-(4-Methylphenyl)-6-[(*Z*)-(4-methylphenyl)-methylidene]-2-phenyl-2-azabicyclo[2.2.2]oct-5-ylidene}aniline**

The title compound, $\text{C}_{34}\text{H}_{32}\text{N}_2$, is formed as a result of the Diels–Alder reaction of *N*-benzylideneaniline and cyclohexen-2-one, catalyzed by zirconium tetrachloride. The molecule exists as the *Z* isomer. All three rings in the azabicyclo[2.2.2]octane system adopt boat conformations. The crystal packing is stabilized by van der Waals forces.

Received 13 June 2005
Accepted 21 June 2005
Online 30 June 2005Imino Diels–Alder
adducts. VII

Comment

The azabicyclo[2.2.2] (isoquinuclidine) ring system has received an increasing amount of attention, mainly due to its occurrence in the natural products dioscorine and ibogaine. In addition, azabicyclo[2.2.2]octanone may serve as a precursor for the synthesis of naturally occurring piperidine alkaloids of the prosopis family (Birkinshaw & Holmes, 1987; Birkinshaw *et al.*, 1988). We have recently synthesized and purified azabicyclo[2.2.2]octane derivatives using an imino Diels–Alder reaction with ZrCl_4 as a 'green' catalyst (Mahesh, 2005). *N*-Benzylideneaniline was reacted with cyclohexen-2-one in the presence of zirconium tetrachloride in dichloromethane at ambient temperature, to afford the title 2-azabicyclo[2.2.2]-octane derivative, (I). The present X-ray structure determination was carried out in order to confirm the identity of (I) and to obtain more detailed information on the conformation of the molecule.



In the molecule of (I), two benzene rings are directly connected to the azabicyclic system and the other two are connected through a double-bond linkage as an extended 'arm' (Fig. 1). In all essential details, the molecular geometry

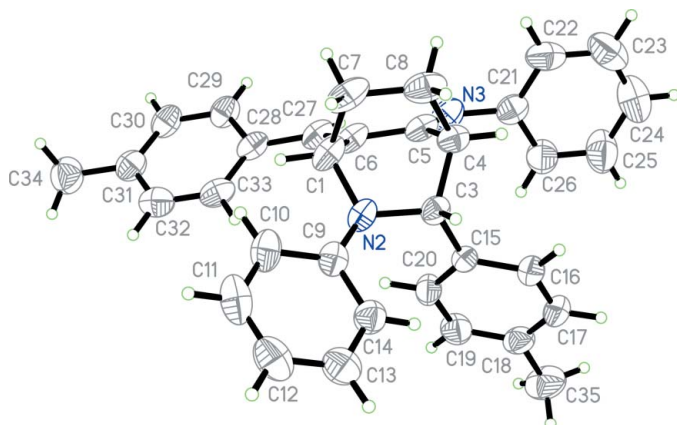


Figure 1
A view of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.

of (I) is not very different from the values found in the literature for other azabicyclo[2.2.2] systems (Sonar *et al.*, 2003, 2004). The disposition of the double bonds C6=C27 and C5=N3 (Table 1) with respect to the azabicyclo[2.2.2] ring system affords the possibility of *E* and *Z* isomers. In the structure of (I), the C27–C28 bond is in a *trans* configuration with respect to the C5–C6 bond [C5–C6–C27–C28 = 175.9 (1)°]. Similarly, the N3–C21 bond is also in a *trans* configuration with respect to the C5–C6 bond [C5–C6–N3–C21 = 175.4 (1)°]. As observed in similar azabicyclo structures, deviations from the ideal bond-angle geometry around the *Csp*² atoms of the double bonds are also found in (I). The C27–C28–C29 [118.8 (2)°], C27–C28–C33 [124.3 (1)°], C5–C6–C27 [121.7 (2)°], N3–C21–C22 [122.9 (2)°], N3–C21–C26 [118.1 (2)°] and N3–C5–C6 [121.1 (1)°] bond angles are close to the ideal value of 120°, while the C1–C6–C27 [127.5 (2)°] and N3–C5–C4 [128.3 (2)°] angles are more distorted, as a consequence of the strain induced by the C6=C27 and C5=N3 double-bond linkages. The sum of the angles around atom N2 is 351.2° and the deviation of atom N2 from the C1/C3/C9 plane is 0.234 (1) Å, indicating its pyramidal nature.

In the azabicyclo[2.2.2] system, all three six-membered rings adopt boat conformations, with asymmetry parameters (Nardelli, 1983) $\Delta C_s(C1)$ of 0.019 (1), 0.023 (1) and 0.040 (1) for the rings C3/N2/C1/C7/C8/C4, C1/C4–C8 and C1/N1/C3–C6, respectively. The torsion angles C6–C27–C28–C33 and C5–N3–C21–C22 are –37.1 (2) and –80.8 (2)°, respectively. The benzene ring substituted at N2 is almost coplanar with the nearest arm of the azabicyclo[2.2.2] ring system, with a C1–N2–C9–C10 torsion angle of 7.6 (2)°, while the benzene ring attached at atom C3 is twisted, with a C4–C3–C15–C16 torsion angle of 72.2 (2)°.

In the absence of H-donating groups, the crystal packing (Fig. 2) is stabilized by van der Waals forces.

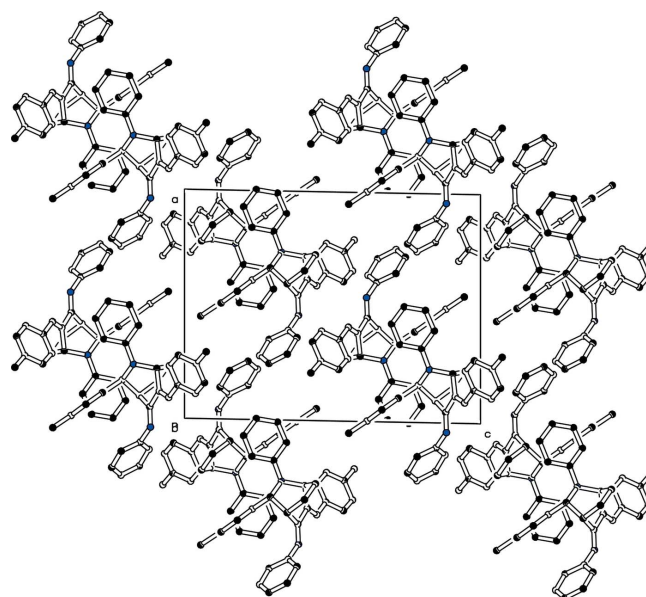


Figure 2
A packing diagram for (I), viewed down the *c* axis. H atoms have been omitted.

Experimental

To a solution of *N*-benzylideneaniline (5.5 mmol) in CH₂Cl₂ (5 ml) at room temperature, ZrCl₄ (10 mol%) and cyclohexen-2-one (5.5 mmol) were added sequentially and the mixture was stirred for 6 h. After completion of the reaction, as indicated by thin-layer chromatography, the reaction was quenched with water, diluted with CH₂Cl₂ (20 ml) and washed with water (10 ml). The aqueous layer was extracted with CH₂Cl₂ (2 × 10 ml). The combined organic layers were dried over Na₂SO₄ and concentrated *in vacuo* to yield a dark-brown residue, which was purified by column chromatography using 2–10% ethyl acetate in hexane as eluent, to obtain the pure product.

Crystal data

C₃₄H₃₂N₂
M_r = 468.62
 Monoclinic, *P*2₁/*n*
a = 14.1805 (8) Å
b = 10.1822 (5) Å
c = 18.3636 (10) Å
 β = 91.080 (1)°
V = 2651.0 (2) Å³
Z = 4

D_x = 1.174 Mg m^{–3}
 Mo K α radiation
 Cell parameters from 6129 reflections
 θ = 2.2–24.2°
 μ = 0.07 mm^{–1}
T = 273 (2) K
 Block, colourless
 0.20 × 0.15 × 0.10 mm

Data collection

Bruker SMART APEX CCD area-detector diffractometer
 ω scans
 Absorption correction: none
 18613 measured reflections
 4659 independent reflections

3617 reflections with $I > 2\sigma(I)$
 R_{int} = 0.019
 θ_{max} = 25.0°
 h = –16 → 16
 k = –12 → 12
 l = –21 → 21

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)]$ = 0.043
 $wR(F^2)$ = 0.127
 S = 1.03
 4659 reflections
 357 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0649P)^2 + 0.3061P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} < 0.001$
 $\Delta\rho_{max} = 0.15 \text{ e \AA}^{-3}$
 $\Delta\rho_{min} = -0.14 \text{ e \AA}^{-3}$

Table 1
Selected bond lengths (Å).

C1–N2	1.4770 (19)	C6–C27	1.337 (2)
C5–N3	1.272 (2)	C29–C30	1.373 (2)

H atoms were included in calculated positions and refined as riding, with C–H distances in the range 0.93–0.98 Å and with $U_{\text{iso}}(\text{H}) = 1.2\text{--}1.5U_{\text{eq}}$ of the parent atom.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINTE* (Bruker, 2001); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

MM thanks the Council of Scientific and Industrial Research (CSIR), India, for the award of a Senior Research Fellowship.

References

- Birkinshaw, T. N. & Holmes, A. B. (1987). *Tetrahedron Lett.* **28**, 813–816.
- Birkinshaw, T. N., Tabor, A. B., Holmes, A. B., Kaye, P., Mayne, P. M. & Raithby, P. R. (1988). *J. Chem. Soc. Chem. Commun.* pp. 1599–1600.
- Bruker (2001). *SMART* (Version 5.625) and *SAINTE* (Version 6.28a) Bruker AXS Inc. Madison, Wisconsin, USA.
- Mahesh, M. (2005). PhD thesis, Indian Institute of Chemical Technology, Hyderabad, India.
- Nardelli, M. (1983). *Acta Cryst.* **C39**, 1141–1142.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Sonar, V. N., Parkin, S. & Crooks, P. A. (2003). *Acta Cryst.* **E59**, o2010–o2012.
- Sonar, V. N., Parkin, S. & Crooks, P. A. (2004). *Acta Cryst.* **C60**, o659–o661.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.